SOLID PHASE SYNTHESIS OF NONADECATHYMIDYLIC ACID BY THE PHOSPHOTRIESTER APPROACH

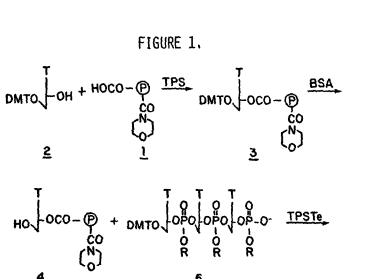
> Ken-ichi Miyoshi and Keiichi Itakura* City of Hope National Medical Center Division of Biology Duarte, California 91010 USA

ABSTRACT Nonadecathymidylic acid was synthesized by the block coupling phosphotriester approach on a polymer support.

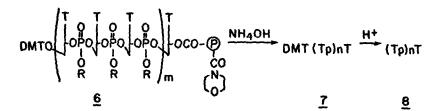
Solid phase synthesis of oligonucleotides has not been as successful as that of peptide synthesis. There may be three reasons for this relative lack of success. 1) The attention has focused almost entirely on the phosphodiester method¹, with its inherent disadvantages, such as low yields in coupling reaction. 2) The commonly used supports have been polystyrene and its derivatives. These non-polar polymers do not swell very well in the polar solvents required for oligonucleotide synthesis², and 3) the stepwise coupling approach³ gives a series of oligonucleotides with one nucleotide difference and, therefore, it is very difficult to isolate the desired product from the final product mixture.

In this communication, we wish to report the synthesis of nonadecathymidylic acid by the block coupling phosphotriester approach on a well swollen polymer support. The polymer <u>1</u> (1 g, 1 mmole equivalent of carboxyl group), the partially hydrolyzed derivative of cross linked polyacrylmorpholide (Enzacryl Gel K2), was shaken with 5'-O-dimethoxytritylthymidine (<u>2</u>, 4 mmoles) and 2,4,6triisopropylbenzenesulfonyl chloride (8 mmoles) in a mixture of acetonitrile and pyridine (1:1 $V/_V$) for 24 hours at room temperature. The unreacted activated carboxyl groups were blocked by shaking with an excess of methanol. The dimethoxytrityl group of the resulting polymer <u>3</u> was removed by treatment with a 2% solution of benzenesulfonic acid in CHCl₃-MeOH (7:3 $V/_V$)⁴ to give the polymer <u>4</u> containing 0.140 mmole of thymidine. For the extension of oligonucleotide chain, the trithymidylic acid block <u>5</u> (5 equivalent) was coupled to the polymer <u>4</u> with the aid of a coupling reagent, 2,4,6-triisopropylbenzenesulfonyl tetrazolide⁴ (TPSTe, 10 equivalent) for 3 hours. The polymer was washed with pyridine and, subsequently, with methanol.

3635



- 0 --\$-x



DMT = dimethoxytrityl T= thymidine BSA = benzenesulfonic acid TPS (x=-CI) TPSTe $(X = -N \xrightarrow{N=N}_{C=N})$ R = - ()- CI

After the removal of the dimethoxytrityl group from the oligothymidyl poly mer $\underline{6}$, the coupling cycle was repeated (see Table 1).⁵ After each coupling cycle, a sample of the polymer 6 (m = 1,2,3,4,5 and 6) was subjected to hydroly

TABLE 1.								
STEP	REAGENT OR SOLVENT	MINUTES SHAKING	NUMBER OF OPERATIONS					
1	2% BSA in CHCl ₃ -MeOH (7:3 V/V)	5	3					
2	CHCl ₃ -MeOH (7:3 ^V / _V)	2	3					
3	Pyridine	2	3					
4	Trinucleotide and TPSTe in pyridine	180	1					
5	Pyridine	2	3					
6	MeCH	2	3					

sis with concentrated ammonium hydroxide solution for 24 hours at 50°C to remove the p-chlorophenyl protecting groups and also to cleave the oligonucleotides from the polymer. The reaction mixture was filtered and the filtrate was concentrated and treated with 80% aqueous acetic acid for 10 minutes at room temperature to give each oligothymidylic acid, $\underline{8}$, (n = 3, 6, 9, 12, 15 and 18). The isolation and analysis of each product was carried out by high performance liquid chromatography (HPLC) on Permaphase AAX.⁶ The yield of each oligomer is shown in Table 2, and the HPLC pattern of the final product, dT_{10} , is presented in Figure 2.

TABLE 2.

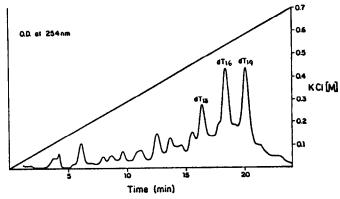
The conversion yields are estimated by comparison of the molar ratios of the product peak (chain length n) to the peak corresponding to chain length n-3 after the separation of the product mixture by HPLC.

Oligonucleotide <u>8</u>	dT ₄	dT7	dT10	dT 13	dT16	dT 19
Yield (%)	77	75	79	71	58	48

The purity of each oligomer was analyzed by gel electrophoresis on 20% polyacrylamide gel after incubation of each oligomer with $[\gamma - {}^{32}P]$ -adenosine 5'triphosphate in the presence of T₄ polynucleotide kinase.⁷ One major labelled product was obtained from each substrate. Finally, the identity of nonadecathymidylic acid was established by partial digestion of the labelled product with venom phosphodiesterase which showed nineteen spots on homochromatography⁸.

FIGURE 2.

The chromatography was performed on a Permaphase AAX column with a linear gradient of KCl as published $^{10}\hdots$



This study shows that the phosphotriester method to construct oligothymidylic acid on the polymer support $\underline{1}$ is simple and rapid. The complete cycle for the addition of one trinucleotide block can be accomplished in less than five hours. Although each coupling reaction did not go to completion, the major advantage of our approach using a trinucleotide block instead of a mononucleotide as an incoming component lies in the isolation of the final product. The desired products are easier to separate from the by-products lacking a trinucleotide block than those from the by-products lacking only a mononucleotide residue. With the improvement of the rapid synthesis of trinucleotide block by the phosphotriester approach⁹, extension of the methodology to the synthesis of oligonucleotides containing all four bases is currently under investigation.

Acknowledgements

This work is supported by USPHS grant to K. Itakura (GM 24293). K. Miyoshi is supported by a predoctoral training grant from Wakunaga Pharmaceutical Co.

References

 Although early examples of oligonucleotide synthesis on a polymer support by the phosphotriester approach were reported, most of the recent work has utilized the phosphodiester method. For reviews on solid phase oligonucleotide synthesis, see:

V. Amarnath and A.D. Broom Chem. Rev. <u>77</u>:183 (1977).

H. Kössel and H. Seliger Fortsuchr. Chem. Org. Naturstoffe <u>32</u>:297 (1975).

2. Recently polymers suitable for use in conjunction with polar reaction media required for oligonucleotide synthesis have been developed and used for the synthesis of oligonucleotides by the phosphodiester approach. These polar polymers have been proven to be much better supports than non-polar polymers, and higher yields of coupling reactions on these polar polymers have been reported:

M.J. Gait and R.C. Sheppard Nucleic Acids Res. <u>4</u>:1153 and 4391 (1977). C.K. Narang, K. Brunfeldt and K.E. Norris Tetrahedron Letters 1819 (1977).

- 3. Adding stepwise one single nucleotide to the elongated chain.
- Y. Stawinski, T. Hozumi, S.A. Narang, C.P. Bahl and R. Wu Nucleic Acids Res. 4:353 (1977).
- Masking of the unreacted 5'-hydroxyl group from the previous coupling cycle was not attempted.
- 6. R.A. Henry, J.A. Schmidt and R.C. Williams J. Chrom. Sci. <u>11</u>:358 (1973).
- 7. C.C. Richardson Proc. Nat. Acad. Sci. USA 54:158 (1965).
- E. Jay, R. Bambara, P. Padmanabhan and R. Wu Nucleic Acids Res. <u>1</u>:331 (1974).
- 9. T. Hirose, R. Crea and K. Itakura Tetrahedron Letters 2449 (1978).
- R. Crea, A. Kraszewski, T. Hirose and K. Itakura Proc. Nat. Acad. Sci. USA <u>75</u>:5765 (1978).

(Received in USA 16 May 1979)